MVP. Samaj's

K.G.D. M. Arts Commerce and Science College, Niphad . Department Of Chemistry

Courses that include experiential learning through project work/field work/internship 1. T.Y.B.Sc. Chemistry Syllabus

Savitribai Phule Pune University, Pune T.Y.B.Sc. Chemistry Syllabus

To be implemented from June 2015 (Academic Year 2015-16) Preamble of the Course

- 1. T.Y.B.Sc. Chemistry is consisting of six theory and three practical courses.
- Each theory course is of 48 lectures; 4 lectures per course per week should be conducted in every semester.
- 3. Out of five optional courses recommended for CH-336 and CH-346, only one option should be taught and the same course should be implemented for the next semester.
- 4. Each practical course is of 4 lectures per week per batch. Practical batch for each course should comprise of 12 students only.
- Each theory paper will carry 50 Marks out of which 10 Marks will be allotted for Internal assessment and University Examination will be conducted for 40 Marks at the end of each semester.
- The practical examination of six hours for each practical course will be conducted at the end of Semester-IV. Each practical course will carry 100 Marks out of which 20 Marks will be allotted for Internal assessment and University Examination will be conducted for 80 Marks.
- 7. Marks for internal assessment of Practical courses will be allotted as follows.
 - a. Completed and Certified journal and regularity of the student

10 Marks

b. Oral Examination and Internal Test

10 Marks

- 8. Internal assessment for theory courses will be done on the basis of the performance of the student in tests. Minimum two tests should be arranged for each course in a Semester.
- 9. Visit to a chemical industry may be organized during the academic year.

NIPHAD IN (NASHIK)

Principal

Karmaveer Ganpat Dada More
Arts.Commerce & Science College.

Niphad Dist.Nashik.

CH-348 - INORGANIC CHEMISTRY PRACTICALS

A) Gravimetric estimations (Any 3)

- 1. Fe as Fe₂O₃
- 2. Nickel as Ni DMG
- 3. Al as Aluminum oxide
- 4. Gravimetric estimation of Ba as BaSO₄ using homogeneous precipitation method.

B) Volumetric Estimations (Any 4)

- 1. Mn by Volhard's method
- 2. Estimation of NO2 by using KMnO4.
- 3. Estimation of % purity of given sample of Sodium Chloride
- 4. Analysis of Brass-Estimation of copper by lodometry
- 5. Fertilizer analysis (PO₄³⁻)

C) Inorganic preparations (Any 4)

- Preparation of Hexamminenickel(II), [Ni (NH₃)₆]²⁺.
- Preparation of Potassium Trioxalatoferrate (III), K₃[Fe(C₂O₄)₃].
- 3. Preparation of Tetraamminecopper (II) suplhate, [Cu (NH₃)₄] SO₄.
- 4. Preparation of Manganese (III) acetylacetonate [Mn(acac)3].
- 5. Preparation of Tris(Thiourea)Copper (I) Chloride [Cu (Thiourea)3]Cl.

D) Colorimetric Estimations (Any 2)

- Iron by thiocyanate method.
- 2. Cobalt by using R-nitroso salt method.
- Titanium by H₂O₂.

E) Separation of binary mixture of cations by Column Chromatography (3 mixtures)

(One mixture should be colorless, Zn + Al, Zn + Mg)

OR

E) Flame Photometry (Any 3)

- 1. Estimation of Na by flame photometry by calibration curve method.
- 2. Estimation of Na by flame photometry by regression method.
- 3. Estimation of K by flame photometry by calibration curve method.
- 4. Estimation of K by flame photometry by regression method.

F) Qualitative Analysis (4 mixtures including Borates and Phosphates)

G) Visit to a chemical industry and report writing is compulsory.

Reference Books: Ref. 1 General Chemistry Experiment – Anil J Elias (University press).

- Ref. 2 Vogel Textbook of Quantitative Chemical Analysis G.H. Jeffery, J. Basset.
- Ref. 3 Quantitative Chemical Analysis S. Sahay (S. Chand & Co.).
- Ref. 4 Quantitative Analysis R.A. Day, Underwood (Prentice Hall).
- Ref. 5 Practical Chemistry K.K. Sharma, D. S. Sharma (Vikas Publication).
- Ref. 6 Vogel's Textbook of Quantitative Chemical Analysis.
- Ref. 7 Monograph on Green Chemistry Laboratory Experiments by Green Chemistry Task Force Committee, DST.

Ref. 8"Experimental Methods in Inorganic Chemistry." Tanaka, J. and Suib, S.L., Prentice Hall, New Jersey, 1999.



Principal

Karmaveer Canpat Dada More

Arts, Commerce & Science College

Niphad Dist, Nashik.

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M.V.P.Samaj's K.G.D.M. Art's, Commerce & Science College, Niphad (Nashik)

Department of Chemistry

Academic Year: - 2019-2020

A Report

On

"Study tour to IIT Bombay SAIF Department"

For B.Sc Chemistry Students



Indian Institute of Technology Bombay (IIT Bombay)

(Date-09/01/2020)

Head
Department of Chemistry
G.D.M. Arts, Commerce & Science Colle 1.
Ninhad, Dist. Nashik



Principal
Karmveer Ganpat Dada More
Arts. Commerce & Science College
Nipnad, Tal.Niphad, Dist.Nashik

Preamble:

K.G.D.M. Art's, Commerce & Science College, Niphad (Nashik) organized a one day Educational Tour to Indian Institute of Technology Bombay (IIT Bombay) on 09/01/2020 to for B.Sc Chemistry students. The visit was organized with the prior permission (IIT Saif department) Teacher specially Prof. B. B. Muntode, Prof. Y. D. Kadlag, Prof. S. S. Kushare, Smt. B. L. Ugale and Smt. P.A. Shinde have taken hard efforts and initiative under the continuous guidance of our Principal Dr. R. N. Bhavare, and Tour in-charge Prof. S. S. Kushare, which made this visit a grand success. Total 45 students along with 4 faculty member and 1 non-teaching assistant Mr. N.B. Dhomase have joined this study visit.

Objective:

- 1) To acquire basic knowledge and functioning of sophisticated analytical instrument.
- 2) To give students an opportunity to relate the classroom learning to the real world situation.
- The student will learn the laboratory skills needed to design, safely conduct and interpret chemical research.
- 4) The student will understand the importance of the analytical instruments, how it works and its role in chemical research.
- 5) Acquire an ability to think scientifically, independently and to make rational discussion.
- 6) Develop an appreciation of chemistry and its application in daily life.
- 7) To teach students to analyze data from experiments or from other sources
- 8) To provide students with some insight into future career prospect in the fields related to Chemistry.



Detail report:

M.V.P.Samaj's K.G.D.M. Art's, Commerce & Science College, Niphad department of chemistry organised one day study tour for B.Sc Chemistry students on 9th January 2020. Study tour organised at Indian Institute of Technology Bombay (IIT Bombay), Department of Sophisticated analytical instrument facility and Centre for research in Nano-technology and Science. Visit starts from Niphad at Thursday 05.00 am and reached to Indian Institute of Technology Bombay (IIT Bombay) at 10.15 am. First we visit to IIT campus and take tea and snacks, after this we start the visit to Sophisticated analytical instrument facility department and visit one by one to different sophisticated instruments. Technical assistants of every sophisticated instrument give us information about principle, working and applications of sophisticated instruments.

During the visit we observe following instruments:

Research Area Wise Facilities available at SAIF Centres.

Elemental Analysis:

- 1) CHNSO Elemental Analyzer(CHNS)
- 2) ICP-Atomic Emission Spectrometer(ICP-AES)
- 3) ICP-Mass Spectrometry(ICP-MS)

Speciation/free radical analysis:

- 1) Electron Spin Resonance Spectrometer(ESR)
- 2) Nuclear Magnetic Resonance Spectrometer (NMR 600 MHz)

Electron Microscopes:

- 1) Environmental Scanning Electron Microscope (ESEM) (Central facility at CRNTS)
- 2) Field Emission Gun-Scanning Electron Microscopes(FEG-SEM)



- 3) Transmission Electron Microscope(TEM)
- 4) High resolution Transmission Electron Microscope 200 kV (HR-TEM 200 kV)(Central facility at CRNTS)
- 5) High Resolution-Transmission Electron Microscope 300 kV (HR-TEM)

Optical Spectroscopy:

- 1) FTIR-Imaging System (FTIR-IMG)
- 2) Laser Raman Spectroscopy (LRS) (Central facility at CRNTS)
- 3) Gas and Liquid Chromatography:
- 4) Gas Chromatograph With High Resolution Mass Spectrometer(GC-HRMS)
- 5) High Resolution Liquid Chromatography Mass Spectrometer (HR-LCMS)
- High Resolution Liquid Chromatography Mass Spectrometer (HR-LCMS Orbitrap)
- 7) Liquid Chromatography Mass Spectrometer (LC-MS)

Other Instruments at SAIF/CRNTS:

- 1) Time-of-Flight Secondary Ion Mass Spectrometer (ToF SIMS)
- Small Angle X-ray Scattering (SAXS).

Return to College:

Return to college at the way visit the historical place in Mumbai at Gateway of India, Build in 1924 George Wittet had prepared a drawing of gateway way of india, which was built in the memory of the visit of king George Vth and Queen Mary to india in 1911, we reached at College on 09th January 2020 at 11.57 pm.

All faculty member and students are thankful to our Respected Principal Dr. R. N. Bhavare for granting the permission and valuable guidance for successful organising study tour.

Department of Chemistry G.D.M. Arts. Commerce & Science Colle 1

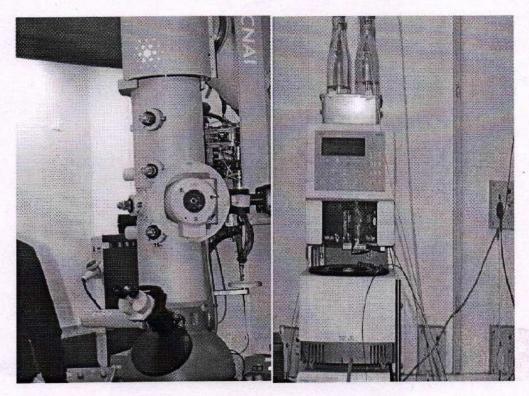
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Karmveer Ganpat Dada More Arts, Commerce & Science College Niphad, Tal. Niphad, Dist. Nashik

Niphad Nashik

Photograph during Visit:



1) Transmission Electron Microscope (TEM)

2) HPLC instrument



3) Prof. B. B. Muntode with Technical assistant at SAIF IIT department.



4) Group Photo Teacher with students at SAIF Department IIT.



5) Group photo teacher with student at Gateway of India.

Department of Chemistry

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Marmveer Ganpat Dada More Arts, Commerce & Science College Niphad, Tal.Niphad, Dist.Nashik

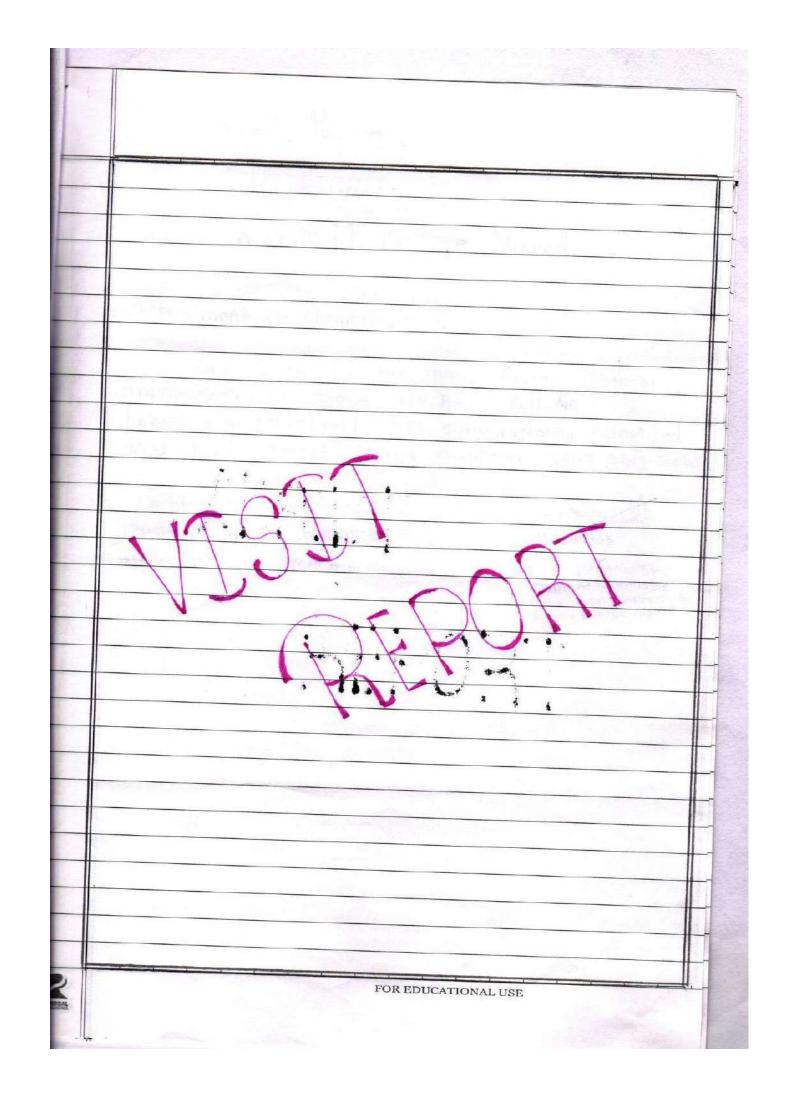
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VISIT REPORT



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Savitaibai Phwe pune university and our college chemistay department of s.4.BSC and T.4.BSC arrange on one day program of visit. IIt Bombay situated in pawai in which we are visit sout Department [spo sophisticated Analytical instauement facility] we was visit salf at date of 9 JAN 2020 (Thueoday) we are visited sAIF department in order to study different microscope spectroscopy and chromotograph techniques. on 9th Jan 2020 early morning 5:30Ar we was started one tone me mas started a Towney tour with four teacher and s.y. Bsc and T.4.BSC student obout 45, we are visited and IIT bombay compus or about 10:30 cm morning we are started to collect information about out the instaurment their work. preequation should be taken and application with the help of respective teacher for respective instauement. These are many instruement available for study, every reasearch or proffesor gives the information about the respective instrument some instruement we was studied their works constauction wooking, application are as follow @ Electrospin resonance spectrometer. Electron spin resonance spectrocopy is based on the absorption of microwave radiation by on unpaired electron when it is exposed to strong magnetic fixeld FOR EDUCATIONAL USE

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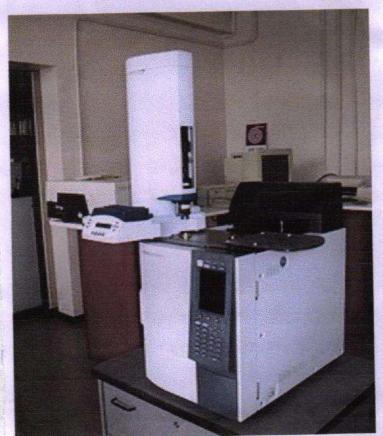
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